

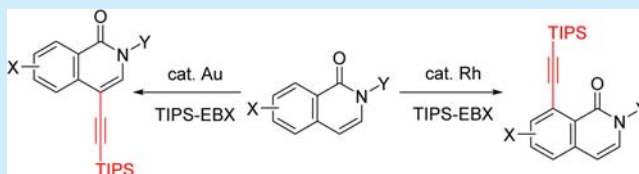
Gold vs Rhodium Catalysis: Tuning Reactivity through Catalyst Control in the C–H Alkynylation of Isoquinolones

Aslam C. Shaikh, Dinesh R. Shinde, and Nitin T. Patil*

Division of Organic Chemistry, CSIR-National Chemical Laboratory, Dr. Homi Bhabha Road, Pune 411008, India
Academy of Scientific and Innovative Research (AcSIR), New Delhi 110025, India

S Supporting Information

ABSTRACT: A site-selective C-4/C-8 alkynylation of isoquinolones catalyzed by gold and rhodium complexes is reported. A broad range of synthetically useful functional groups (–F, –Cl, –Br, –CF₃, –OMe, alkyl, etc.) were tolerated, providing an efficient and robust protocol for the synthesis either C-4- or C-8-alkynylated isoquinolones.



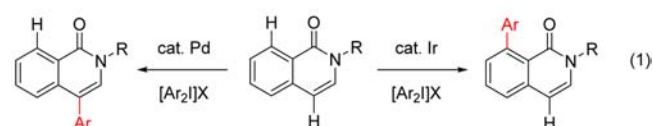
Isoquinolone is a privileged structural motif widely found in natural products and biologically valuable compounds.¹ Consequently, tremendous efforts have been devoted to the effective synthesis and functionalization of isoquinolones. Over the past few decades, transition-metal-catalyzed C–H functionalization has emerged as an attractive, economical, and environmentally benign alternative to the classical synthetic methods, allowing the expeditious formation of diverse isoquinolones.² In particular, the direct C–H functionalization of isoquinolones has witnessed great success for late-stage modification of the preexistent quinolone scaffolds.

Hirano and Miura, for the first time, reported a copper-mediated C6-selective heteroarylation of 2-pyridones in which the observed site selectivity was directed by a pyridyl substituent on the nitrogen atom of the isoquinolone ring.³ Recently, Hong and co-workers reported an efficient catalytic system for the *N*-pyrimidyl group directed C-2 selective C–H alkenylation of 4-quinolones using a decarbonylative coupling strategy.⁴ Very recently, pioneering work from Hong's laboratory revealed an interesting example of catalyst-controlled site selective C–H arylation of isoquinolone using arylodonium salts as the coupling partners (Scheme 1, eq 1).⁵ Recently, the site-selective direct alkynylation of quinolones was reported by them (Scheme 1, eq 2).⁶ The key for the success of this reaction was the use of appropriate group placed at nitrogen atom of quinolones. We were particularly interested in the introduction of an alkyne⁷ in the isoquinolones, as alkynes are a key functional group for many organic transformations. Herein, we describe a site-selective C-4/C-8 alkynylation of isoquinolones enabled by gold and rhodium complexes (Scheme 1, eq 2). Based on Hong's report (Scheme 1, eq 1)⁵ and our continued interest in the exploration of molecular diversity through catalyst control,⁸ we speculated that C-4 selectivity can be achieved with the use of electrophilic metal catalysts while the C-8 selectivity is expected when metal catalyst capable of coordinating the carbonyl group causing C8–H activation is employed.

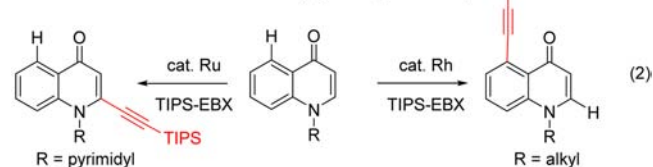
Our initial investigation focused on the reaction of *N*-methylisoquinolone (1a) and 1-[(triisopropylsilyl)ethynyl]-1,2-

Scheme 1. Site-Selective C–H Functionalization of Isoquinolones: Known and Present Work

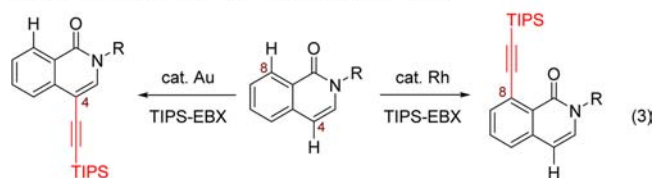
Known Work: Site selective C–H arylation (Ir Vs. Pd)



Known Work: Site selective C–H alkynylation (Ru Vs. Rh)



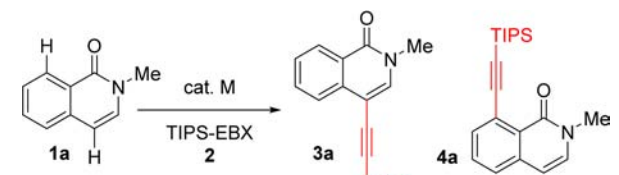
This Work: Site selective C–H alkynylation (Au Vs. Rh)



benziodoxol-3(1*H*)-one (TIPS-EBX) (2), which was first reported by Waser in the alkynylation of indoles with gold or palladium catalysis⁹ and further applied to Rh(III)- or Ir(III)-catalyzed C–H alkynylation by Li,¹⁰ Loh,¹¹ and Glorius.¹² When *N*-methylisoquinolone (1a) was treated with TIPS-EBX in the presence of AuCl₃/AuBr in DCE at 25 °C, the desired C4-alkynylation product 3a was not obtained at all; starting material was recovered in a quantitative amount (Table 1, entries 1 and 2). Pleasingly, 3a was obtained in 54% yield when AuCl was used as catalyst (entry 3). The reaction worked with a broad range of solvents (entries 3–6); however, more promising results were obtained in the case of dry CH₃CN. The product 3a was isolated

Received: January 19, 2016

Published: February 17, 2016

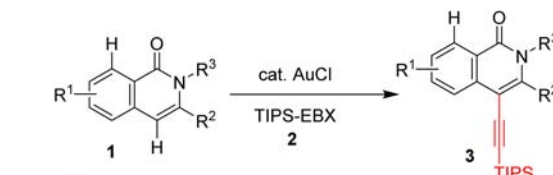
Table 1. Optimization Studies^a


entry	cat.	additives	solvent	temp (°C)	yield ^b (%)	
					3a	4a
1	AuCl ₃		DCE	25	trace	
2	AuBr		DCE	25		
3	AuCl		DCE	25	54	
4	AuCl		Et ₂ O	25	64	
5	AuCl		CHCl ₃	25	74	
6	AuCl		CH ₃ CN	25	81	
7	AuCl		CH ₃ CN	50	92	
8	[RhCp*Cl ₂] ₂		DCE	25		trace
9	[RhCp*Cl ₂] ₂	Cu(OAc) ₂	DCE	25		trace
10	[RhCp*Cl ₂] ₂	Zn(OTf) ₂	DCE	25		trace
11	[RhCp*Cl ₂] ₂	AgOAc	DCE	25		13
12	[RhCp*Cl ₂] ₂	AgSbF ₆	DCE	25		44
13	[RhCp*Cl ₂] ₂	AgSbF ₆	DCE	50		62
14	[RhCp*Cl ₂] ₂	AgSbF ₆	DCE	80		88 ^c

^aReaction conditions: (1) 0.15 mmol of **1a**, 0.18 mmol of **2**, 10 mol % of catalyst, dry solvent (2.0 mL), 24 h (entries 1–7); (2) 0.15 mmol of **1a**, 0.18 mmol of **2**, 2.5 mol % of [RhCp*Cl₂]₂, 5 mol % of additives, dry DCE (2.0 mL), 16 h (entries 8–14). ^bIsolated yields. ^c10 mol % of AgSbF₆ was used.

in 92% yield at 50 °C when reaction was run for 24 h. Interestingly, in none of the cases was C8-alkynylation product **4a**. In the search for a catalytic system suitable for obtaining **4a**, we screened various catalysts capable of affecting chelation-assisted C–H activations. Although there exist a number of reports on the C–H functionalization using heteroatom as directing groups, very few reports exist where carbonyl (or amide carbonyl) groups have been used as directing groups.¹³ Hence, rhodium catalysts in combination with metal salts have been examined at 25 °C (entries 8–10); however, the formation of **4a** was not observed. Interestingly, AgOAc, a popular additive for [RhCp*Cl₂]₂ catalyzed C–H activation reactions, gave **4a**, albeit in 13% yield (entry 11). Switching the additive from AgOAc to AgSbF₆ increased the yield of **4a** up to 44% (entry 12). With an increase in the temperature up to 50 °C, **4a** was obtained in 62% yield (entry 13). A significant improvement in yield was noticed when the amount of AgSbF₆ was increased up to 10 mol % and the temperature up to 80 °C (entry 14, 88%).

With the optimized reaction conditions in hand, the scope of the reaction was examined with various isoquinolones. As shown in Table 2, C-4 selective alkynylation of isoquinolones proceeded well to give the desired products irrespective of substitution patterns on the phenyl ring. Both electron-donating and -withdrawing substituents, such as methyl (**3f** and **3g**), methoxy (**3h** and **3i**), fluoro (**3j**), chloro (**3k**), bromo (**3l**), and trifluoromethyl (**3m**), were compatible, providing alkynylisoquinolones in moderate to good yields. It should be noted that *N*-phenyl-substituted isoquinolones were unsuccessful in yielding the desired coupling products under the present reaction conditions (**3e**) probably because of the lower nucleophilicity. Interestingly, introduction of a substitution at C-2 position of isoquinolones did not hamper the outcome of the reaction, and

Table 2. Gold-Catalyzed C-4 Alkynylation^a


entry	cat.	additives	solvent	temp (°C)	yield (%)
1	AuCl		DCE	25	trace
2	AuBr		DCE	25	
3	AuCl		DCE	25	54
4	AuCl		Et ₂ O	25	64
5	AuCl		CHCl ₃	25	74
6	AuCl		CH ₃ CN	25	81
7	AuCl		CH ₃ CN	50	92
8	[RhCp*Cl ₂] ₂		DCE	25	trace
9	[RhCp*Cl ₂] ₂	Cu(OAc) ₂	DCE	25	trace
10	[RhCp*Cl ₂] ₂	Zn(OTf) ₂	DCE	25	trace
11	[RhCp*Cl ₂] ₂	AgOAc	DCE	25	13
12	[RhCp*Cl ₂] ₂	AgSbF ₆	DCE	25	44
13	[RhCp*Cl ₂] ₂	AgSbF ₆	DCE	50	62
14	[RhCp*Cl ₂] ₂	AgSbF ₆	DCE	80	88 ^c

3f, R¹ = Me, R³ = Me, 90 %
3g, R¹ = Me, R³ = ⁱPr, 86 %
3h, R¹ = MeO, R³ = Me, 88 %
3i, R¹ = MeO, R³ = Cyclopentyl, 85 %
3j, R¹ = F, 40 %
3k, R¹ = Cl, 42 %
3l, R¹ = Br, 64 %
3m, R¹ = CF₃, 58 %
3a, R³ = Me, 92 %
3b, R³ = Et, 88 %
3c, R³ = ⁿBu, 86 %
3d, R³ = CH₂Ph, 82 %
3e, R³ = Ph, -%^b
3n, R² = Cyclopropyl, 90 %
3o, R² = ⁿBu, 85 %
3p, R² = Ph, 88 %
3q, 76 %

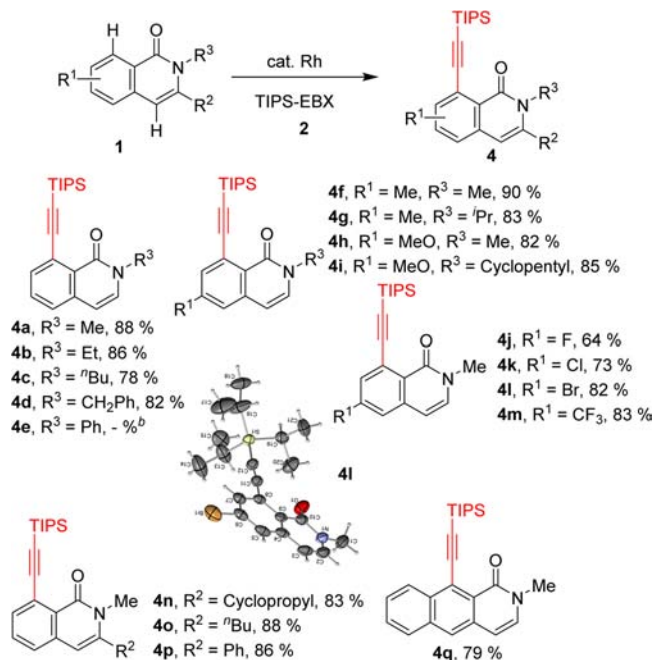
^aReaction conditions: 0.15 mmol of **1a**, 0.18 mmol of **2**, 10 mol % of AuCl, dry CH₃CN (2.0 mL), 50 °C, 24 h. ^bNo product formation, starting material recovered. ^cIsolated yields are given.

3n–p were obtained in excellent yields (85–90%). In addition, the C–H alkynylation reaction is readily found to proceed with structurally related benzoisoquinolone substrate to give the desired alkynylation product **3q** in 76% yield.

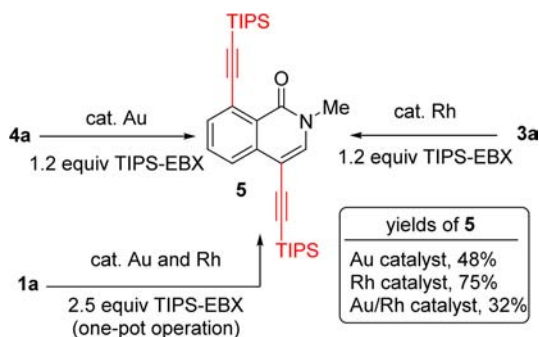
We next examined the scope of the C-8-selective alkynylation reaction. As shown in Table 3, isoquinolone bearing halogen as well as electron-donating and -withdrawing groups in the phenyl ring underwent smooth reaction with TIPS-EBX (**2**) to obtain the corresponding C-8 alkynylation products in good yields. The C-2-substituted isoquinolones were also found to be good substrates affording the corresponding products in good yields (**4n–p**, 85–90%). Similarly, when the benzoisoquinolone substrate was submitted to the optimal reaction conditions, the product **4q** was isolated in 79% yield.

Next, we endeavored to achieve C-4/C-8 dialkynylation in a stepwise as well as in a one-step fashion (Scheme 2). Thus, when **3a** was treated with standard rhodium catalysis, product **5** was obtained in 75% yield. Similarly, the treatment of **4a** with standard gold catalysis conditions afforded **5** in 48% yield. Interestingly, dialkynylation product **5** could also be obtained in a one-pot fashion under Au/Rh relay catalysis,¹⁴ albeit in 32% yield.

On the basis of the Au-mediated alkynylation of indole, as reported by Waser^{9d} and proven theoretically by Ariafard,¹⁵ the proposed mechanism for C-4 selective C–H alkynylation is given in Scheme 3. At first, the coordination of AuCl to the triple bond of TIPS-EBX **2** would take place to form gold–alkyne complex **6**. The gold–alkyne complex **6** would readily convert into vinyl–gold complex **7** via isoquinolone metalation followed by α/β -elimination followed by 1,2-shift depending on the regioselectivity of the addition. Other mechanisms triggered by the oxidative addition of Au(I) within the C–I bond cannot be ruled out.^{9g} Regarding C-8-selective C–H alkynylation, the catalytic cycle could start from [RhCp*Cl₂]₂ (**1**). The first event would be

Table 3. Rhodium-Catalyzed C-8 Alkynylation^a

Scheme 2. Dialkynylation of Isoquinolones

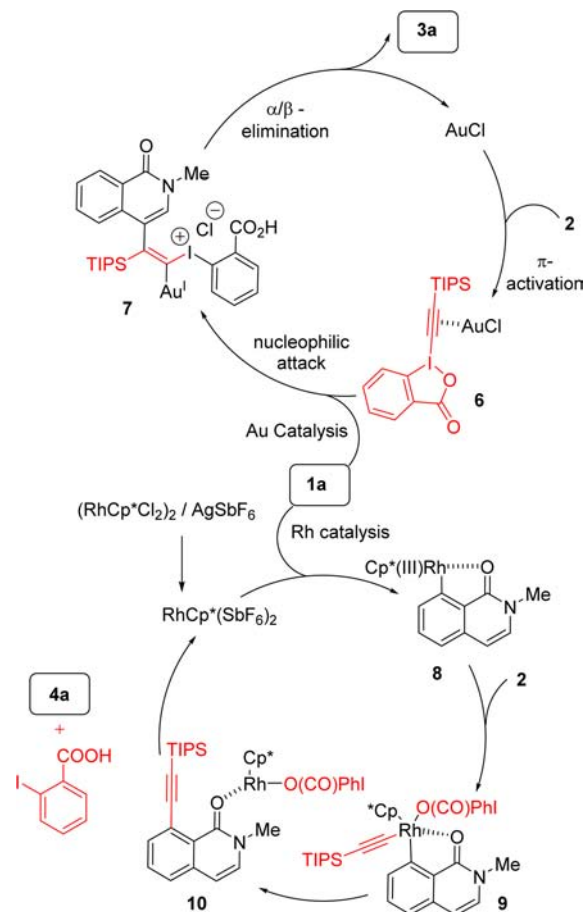


the coordination of catalyst to 1a followed by C–H activation. This process could be the rate-determining step and may follow a concerted metalation/deprotonation pathway leading to intermediate 8 as reported by Glorius and co-workers.¹² Next, the acetylene may coordinate to Rh(III) to yield intermediate 9. From this point, insertion of acetylene in the Rh–C bond can occur to give intermediate 10, which would finally lead to the formation of 4a with the regeneration of Rh catalyst.^{11a}

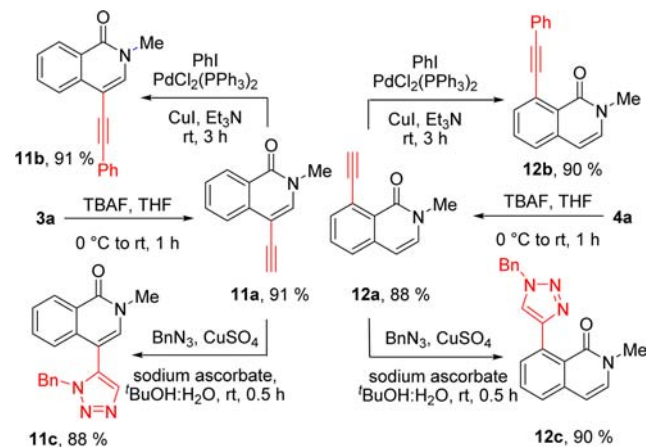
To demonstrate the synthetic utility of alkynylated products, a few interesting organic transformations of 3a and 4a were carried out (Scheme 4). The deprotection of the TIPS group in 3a and 4a was achieved with TBAF to obtain terminal alkynes 11a and 12a in 91 and 88 % yield, respectively. The terminal alkynes 11a/12a were then subjected to Sonogashira reactions and copper-catalyzed azide–alkyne cycloadditions to afford 11b/12b and 11c/12c.

In summary, we developed site-selective (C-4 vs C-8) C–H alkynylation of isoquinolones catalyzed by gold and rhodium catalysts. The silyl-protecting groups could be easily removed to

Scheme 3. Plausible Mechanism



Scheme 4. Utility of Products



give terminal alkynes, a functional handle for many organic transformations.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b00175.

Experimental procedures, analytical data, and ¹H and ¹³C NMR spectra of all newly synthesized products (PDF)

X-ray data for 3n (CIF)

X-ray data for 4l (CIF)

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: n.patil@ncl.res.in.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

Generous financial support by the Department of Science and Technology (DST), New Delhi (Grant No. SB/S1/OC-17/2013) and the Council of Scientific and Industrial Research (CSIR), New Delhi (Grant Nos. CSC0108 and CSC0130) is gratefully acknowledged. A.C.S. thanks CSIR for the award of an SPM Fellowship.

■ REFERENCES

- (1) Reviews: (a) Jin, Z. *Nat. Prod. Rep.* **2013**, *30*, 849–868. (b) Glushkov, V. A.; Shklyav, Y. V. *Chem. Heterocycl. Compd.* **2001**, *37*, 663–687. (c) Lewis, J. R. *Nat. Prod. Rep.* **1994**, *11*, 329–332. For other examples, see: (d) Pettit, G. R.; Ducki, S.; Eastham, S. A.; Melody, N. J. *J. Nat. Prod.* **2009**, *72*, 1279–1282. (e) Khadka, D. B.; Yang, S. H.; Cho, S. H.; Zhao, C.; Cho, W.-J. *Tetrahedron* **2012**, *68*, 250–261. (f) Saeed, A.; Ashraf, Z. *Pharm. Chem. J.* **2008**, *42*, 277–280. (g) Cappelli, A.; Pericot Mohr, G.; Giuliani, G.; Galeazzi, S.; Anzini, M.; Mennuni, L.; Ferrari, F.; Makovec, F.; Kleinrath, E. M.; Langer, T.; Valoti, M.; Giorgi, G.; Vomero, S. *J. Med. Chem.* **2006**, *49*, 6451–6464. (h) Ruchelman, A. L.; Houghton, P. J.; Zhou, N.; Liu, A.; Liu, L. F.; LaVoie, E. J. *J. Med. Chem.* **2005**, *48*, 792–804. (i) Nagarajan, M.; Morrell, A.; Fort, B. C.; Meckley, M. R.; Antony, S.; Kohlhagen, G.; Pommier, Y.; Cushman, M. *J. Med. Chem.* **2004**, *47*, 5651–5661. (j) Coelho, F.; Veronese, D.; Lopes, E. C. S.; Rossi, R. C. *Tetrahedron Lett.* **2003**, *44*, 5731–5735. (k) González, M. C.; Zafra-Polo, M. C.; Blázquez, A.; Serrano, A.; Cortes, D. *J. Nat. Prod.* **1997**, *60*, 108–110.
- (2) (a) Grigorjeva, L.; Daugulis, O. *Angew. Chem., Int. Ed.* **2014**, *53*, 10209–10212. (b) Yu, D.-G.; de Azambuja, F.; Glorius, F. *Angew. Chem., Int. Ed.* **2014**, *53*, 2754–2758. (c) Zhu, W.; Zhang, D.; Yang, N.; Liu, H. *Chem. Commun.* **2014**, *50*, 10634–10636. (d) Reddy, M. C.; Manikandan, R.; Jeganmohan, M. *Chem. Commun.* **2013**, *49*, 6060–6062. (e) Zhong, H.; Yang, D.; Wang, S.; Huang, J. *Chem. Commun.* **2012**, *48*, 3236–3238. (f) Wang, H.; Grohmann, C.; Nimphius, C.; Glorius, F. *J. Am. Chem. Soc.* **2012**, *134*, 19592–19595. (g) Too, P. C.; Chiba, S. *Chem. Commun.* **2012**, *48*, 7634–7636. (h) Ackermann, L.; Lygin, A. V.; Hofmann, N. *Angew. Chem., Int. Ed.* **2011**, *50*, 6379–6382. (i) Lu, J.; Gong, X.; Yang, H.; Fu, H. *Chem. Commun.* **2010**, *46*, 4172–4174. (j) Hyster, T. K.; Rovis, T. *J. Am. Chem. Soc.* **2010**, *132*, 10565–10569. (k) Mochida, S.; Umeda, N.; Hirano, K.; Satoh, T.; Miura, M. *Chem. Lett.* **2010**, *39*, 744–746. (l) Liu, C.-C.; Parthasarathy, K.; Cheng, C.-H. *Org. Lett.* **2010**, *12*, 3518–3521. (m) Wang, F.; Liu, H.; Fu, H.; Jiang, Y.; Zhao, Y. *Org. Lett.* **2009**, *11*, 2469–2472. (n) Zheng, Z.; Alper, H. *Org. Lett.* **2008**, *10*, 4903–4906. (o) Kajita, Y.; Matsubara, S.; Kurahashi, T. *J. Am. Chem. Soc.* **2008**, *130*, 6058–6059. (p) Miura, T.; Yamauchi, M.; Murakami, M. *Org. Lett.* **2008**, *10*, 3085–3088. (r) Batchu, V. R.; Barange, D. K.; Kumar, D.; Sreekanth, B. R.; Vyas, K.; Reddy, E. A.; Pal, M. *Chem. Commun.* **2007**, *19*, 1966–1968.
- (3) Odani, R.; Hirano, K.; Satoh, T.; Miura, M. *Angew. Chem., Int. Ed.* **2014**, *53*, 10784–10788.
- (4) Kwon, S.; Kang, D.; Hong, S. *Eur. J. Org. Chem.* **2015**, *2015*, 3671–3678.
- (5) Lee, S.; Mah, S.; Hong, S. *Org. Lett.* **2015**, *17*, 3864–3867.
- (6) Kang, D.; Hong, S. *Org. Lett.* **2015**, *17*, 1938–1941.
- (7) (a) Landge, V. G.; Shewale, C. H.; Jaiswal, G.; Sahoo, M. K.; Midya, S. P.; Balaraman, E. *Catal. Sci. Technol.* **2016**, DOI: 10.1039/C5CY01299F. (b) Wang, S.; Guo, L.-N.; Wang, H.; Duan, X.-H. *Org. Lett.* **2015**, *17*, 4798–4801. (c) Zhang, Z.-Z.; Liu, B.; Wang, C.-Y.; Shi, B.-F. *Org. Lett.* **2015**, *17*, 4094–4097. (d) Ai, W.; Wu, Y.; Tang, H.; Yang, X.; Yang, Y.; Li, Y.; Zhou, B. *Chem. Commun.* **2015**, *51*, 7871–7874. (e) Shang, M.; Wang, H.-L.; Sun, S.-Z.; Dai, H.-X.; Yu, J.-Q. *J. Am. Chem. Soc.* **2014**, *136*, 11590–11593. (f) He, J.; Wasa, M.; Chan, K. S. L.; Yu, J.-Q. *J. Am. Chem. Soc.* **2013**, *135*, 3387–3390. (g) Ano, Y.; Tobisu, M.; Chatani, N. *J. Am. Chem. Soc.* **2011**, *133*, 12984–12986. (h) Haro, T.; Nevado, C. *J. Am. Chem. Soc.* **2010**, *132*, 1512–1513. (i) Seregin, I. V.; Ryabova, V.; Gevorgyan, V. *J. Am. Chem. Soc.* **2007**, *129*, 7742–7743.
- (8) Bansode, A. H.; Shaikh, A. C.; Kavthe, R. D.; Thorat, S.; Gonnade, R. G.; Patil, N. T. *Chem. - Eur. J.* **2015**, *21*, 2319–2323.
- (9) Review: (a) Brand, J. P.; Waser, J. *Chem. Soc. Rev.* **2012**, *41*, 4165–4179. For other examples, see: (b) Li, Y.; Brand, J. P.; Waser, J. *Angew. Chem., Int. Ed.* **2013**, *52*, 6743–6747. (c) Tolnai, G. L.; Ganss, S.; Brand, J. P.; Waser, J. *Org. Lett.* **2013**, *15*, 112–115. (d) Brand, J. P.; Chevalley, C.; Scopelliti, R.; Waser, J. *Chem. - Eur. J.* **2012**, *18*, 5655–5666. (e) Brand, J. P.; Chevalley, C.; Waser, J. *Beilstein J. Org. Chem.* **2011**, *7*, 565–569. (f) Nicolai, S.; Piemontesi, C.; Waser, J. *Angew. Chem., Int. Ed.* **2011**, *50*, 4680–4683. (g) Brand, J. P.; Charpentier, J.; Waser, J. *Angew. Chem., Int. Ed.* **2009**, *48*, 9346–9349.
- (10) (a) Wang, H.; Xie, F.; Qi, Z.; Li, X. *Org. Lett.* **2015**, *17*, 920–923. (b) Xie, F.; Qi, Z.; Yu, S.; Li, X. *J. Am. Chem. Soc.* **2014**, *136*, 4780–4787. (c) Zhang, X.; Qi, Z.; Gao, J.; Li, X. *Org. Biomol. Chem.* **2014**, *12*, 9329–9332.
- (11) (a) Feng, C.; Loh, T.-P. *Angew. Chem., Int. Ed.* **2014**, *53*, 2722–2726. (b) Feng, C.; Feng, D.; Luo, Y.; Loh, T.-P. *Org. Lett.* **2014**, *16*, 5956–5969. (c) Feng, C.; Feng, D.; Loh, T.-P. *Chem. Commun.* **2014**, *50*, 9865–9868.
- (12) Collins, K.; Lied, F.; Glorius, F. *Chem. Commun.* **2014**, *50*, 4459–4461.
- (13) (a) Huang, Z.; Lim, H. N.; Mo, F.; Young, M. C.; Dong, G. *Chem. Soc. Rev.* **2015**, *44*, 7764–7786. (b) Patureau, F. W.; Besset, T.; Glorius, F. *Angew. Chem., Int. Ed.* **2011**, *50*, 1064–1067. Banerjee, A.; Santra, S. K.; Mohanta, P. R.; Patel, B. K. *Org. Lett.* **2015**, *17*, 5678.
- (14) Patil, N. T.; Shinde, V. S.; Gajula, B. *Org. Biomol. Chem.* **2012**, *10*, 211–224.
- (15) Ariafard, A. *ACS Catal.* **2014**, *4*, 2896–2907.